Effect of Aromatic SAMs Molecules on Graphene/Silicon Schottky Diode Performance

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Graphene is a one-atom thick sheet of sp2 bonded carbon atoms that are arranged in a honeycomb crystal lattice with exceptional properties. At room temperature electron mobility reached 2.5 × 105 cm2/V.s.1 Young’s modulus of 1 TPa and intrinsic strength of 130 GPa are very close to that were predicted by theory.2 Very high thermal conductivity (above 3000 WmK−1),3 low optical absorption (∼2.3%),4 ability to sustain high electric current densities5 and be suitable to chemical functionalization6–8 are other important properties of graphene that attract great attention. When graphene is transferred onto bare and SAM modified n-type silicon substrates, the electrical transport and charge carrier injection.19,20 Additionally, SAM modification of graphene/n-Si Interface properties between substrate and graphene layer.15

Preparation of SAM molecules.—5-[(3-methylphenyl)(phenyl)amino]isophthalic acid (MePIFA) and 5-(diphenyl)amino]isophthalic acid (DPIFA) aromatic small molecules with double bond carboxylic acid were used as self-assembly monolayers (SAMs) (see Figure 1). The synthesis procedure of SAM molecules was reported in our previous study.22 SAM molecules with 1 mM concentration were prepared at room temperature in methanol solution. SiO2(300 nm)/n-Si substrates were kept in methanol-SAM solution for 24 hours to be covered with MePIFA and DPIFA SAM molecules. The substrates were then rinsed with methanol to remove SAM molecules residues. Graphene growth and transfer process.—Cu foil (25 μm thick, 99.8% purity, Alfa Aesar) was used as catalyst-substrate material which on a quartz glass placed in the atmospheric pressure CVD system. First of all, the Cu foil was heated up to 1073°C under H2 (20 sccm) and Ar (200 sccm) with a ramp 30°C min−1 then annealed under same temperature and flows for a one hour. After annealing, CH4 (10 sccm) was introduced in tube furnace for three minutes in order to obtain graphene growth. Finally, sample was left for rapid cooling down to the room temperature in a gas flows of H2 (20 sccm) and Ar (200 sccm).

Microspot S1318 Photoresist (PR) was utilized as supporting layer during the graphene transfer process. We drop casted on the graphene-Cu substrate and waited overnight at 70°C. Iron Chloride (FeCl3) was used for Cu etching. As the Cu foil fully etched away, graphene-PR

Figure 1. Chemical structures of (a) MePIFA and (b) DPIFA SAM molecules.
was cleaned by DI water to remove FeCl₃ residue. After N₂ drying, graphene-PR was transferred to clean SAMs/SiO₂/n-Si (0.01–30 Ω-cm) substrate and bare SiO₂/n-Si substrate. Thereafter, substrates were baked at 110°C in order to provide better adhesion of graphene to the substrates. Finally, PR was removed in hot acetone to yield large area graphene on SAMs/SiO₂/n-Si and SiO₂/n-Si substrate.

Device fabrication and characterization.—We etched a part of SiO₂ to prevent electrical shortening along the graphene layer. Then, transferred graphene placed on etch SiO₂/n-Si substrate. Finally, gold with a thickness of 100 nm was deposited as front and back contact (area = 1 mm) using thermal evaporation technique.

Raman Spectroscopy of transferred graphene was performed using Ar-ion gas laser operating at 488 nm wavelength with 600 groove/mm grating under 100× microscope objective. Sheet resistance of transferred graphene on SiO₂/Si was measured using Van der Pauw’s method. Current-Voltage characteristic of modified SAM molecules SiO₂/n-Si and bare one device were taken with two probe measurement via 2420 SourceMeter under dark condition. Atomic Force Microscopy (AFM) and Kelvin Probe Force Microscopy (KPFM) were performed using TiN cantilever with curvature radius of 35 nm (NTMDT).

Results and Discussion

Surface characterization.—Figure 2a shows difference color contrast in optical image due to inhomogeneity fully coverage graphene films on SiO₂/Si substrate. SEM image of graphene transferred on SiO₂/Si substrate as indicated in Figure 2b. Some wrinkle was observed owing to the difference thermal expansion coefficient between Cu surface and graphene film during the cooling. As seen from Figure 2c, small particles are believed to be SiO₂ were observed in the AFM image, as reported in similar works.²³,²⁴ The formation of SiO₂ can be observed because the Si particles from CVD react with oxygen in APCVD condition. Therefore, they could be probably transferred with graphene on SiO₂/Si.

Raman spectrum indicates typical appearance of transferred graphene on SiO₂/Si including D band (1369 cm⁻¹), G (1587 cm⁻¹) and G’ band (2726 cm⁻¹) respectively (see Figure 3a). The G band is the first Raman peak related to C-C stretching of sp² carbon. The G’ band is the second order graphene peak and the D provides information about the sp³ bonds revealing “amount of disorder” in graphene. In addition to this, Intensity ratio of G/G’ and D/G have been utilized to identify number of graphene layers and defect content of graphene. I_G/G’ and I_D/G were about 0.97 and 0.08 verifying presence of high quality bilayer graphene.²⁵–²⁷

The G’ in bilayer can be deconvoluted in more components due to four possible scattering process as shown in Figure 3b. The bilayer graphene on SiO₂/Si was confirmed by asymmetric G’ band with the full width half maximum (FWHM) of 48 cm⁻¹.²³ Sheet resistance measurement of transferred graphene on SiO₂/n-Si was performed using van der Pauw’s method to investigate quality of the graphene. In this method, while the current was measured on two isolated contacts, the voltage is determined on opposing two contacts.²⁹ Eight measurements were performed around sample and the calculated sheet resistance of graphene was about 298 Ω/sq. This value is in good agreement for CVD grown bilayer graphene as reported in literature.²⁶,³⁰

As a next step, In order to understand modification of n-Si surface with DPIFA and MePIFA SAMs molecules, KPFM measurements were obtained using conductive AFM tip. Surface potential difference (SPD) between tip and sample can be define as follows:

\[ V_{SPD} = \frac{\Psi_{sample} - \Psi_{tip}}{e} \]  

[1]
where ε is the electric charge, ϕ_{sample} and ϕ_{tip} are the work functions of the sample, and TiN tip respectively. According to above equation, it is possible to obtain surface potential in terms of eV defined work functions of bare n-Si and modified n-Si with SAMs molecules. Figure 4 indicates cantilever oscillation amplitude (Mag) as a function of bias voltage. Bare n-Si, n-Si/MePIFA and n-Si/DPIFA were measured as 0.158 V, 0.188 V and 0.383 V respectively. The differences of surface potential reveal the modification of n-Si surface with MePIFA and DPIFA SAMs molecular films. Increasing surface potential is related to enhanced charge density on the n-Si surface. In presence of the SAM molecules, potential distributions were enhanced compare to bare n-Si surface. As a result, surface potentials of SAM modified n-Si surfaces were increased with respect to the bare n-Si surface.

### The current-voltage behavior of schottky diodes

Device structures of bare and modified GRP/n-Si and bonding mechanism between double bond carboxylic acid based SAMs and n-Si with native oxide as illustrated in Figure 5a. Due to formation of ester bond on the surface, charge injection and transport occur through molecules and are enhanced owing to inelastic tunneling mechanism over its π-conjugated structure. Figure 5b indicates schematic energy band diagram of GRP/n-Si junction. The work function differences between n-Si and graphene lead to charge transfer from n-Si to graphene until alignment of Fermi level across the junction and as a result, bending of energy levels near junction take place. In the presence of SAMs molecules as shown in Figure 5c, molecular dipole layer created between n-Si and graphene and charge transfer are facilitated due to reduction of energy barriers and π-bonding between aromatic SAMs molecules and graphene. Under the applied electric field, electron transfer occurs from the donor SAM molecules to n-type silicon, resulting SAM molecules acts as a p-type dopant. Then, holes transfer is facilitated from SAM molecules to graphene due to π-π interaction. As a result, typically p-n junction is created. That’s why we used n-type silicon to form p-n junction.

The electronic parameters of constructed three devices were showed rectifying behavior like metal/semiconductor Schottky contact. The current-voltage characteristics of Au/GRP/n-Si, Au/GRP/MePIFA/n-Si and Au/GRP/DPIFA/n-Si were shown in Figure 6.

The diodes that modified by MePIFA and DPIFA SAM molecules were showed better rectifying behavior when compared to bare graphene based diode. At temperatures above 260 K, current-voltage behavior can be analyzed by thermionic emission as:

\[
I = A \cdot A' T^2 \exp \left( -\frac{q \phi_B}{kT} \right) \exp \left( \frac{q (V - I R_s)}{n kT} \right) - 1 \]

where A is the effective contact area, A' is the Richardson constant, T is the absolute temperature, ϕ_B is the barrier height, \( k \) Boltzmann constant, q is the electronic charge and n is the ideality factor. Due to the value ideality factor was obtained as higher than unity, the effect of series resistance (R_s) should be included and relation can be rewritten as:

\[
I = A \cdot A' T^2 \exp \left( -\frac{q \phi_B}{kT} \right) \exp \left( \frac{q (V - I R_s)}{n kT} \right) \]

\[ q (V - I R_s) > kT \]

If Eq. 3 is rearranged:

\[ V = n \frac{kT}{q} \ln \left( \frac{I}{A \cdot A' T^2} \right) + I R_s + n \phi_B \]

When Eq. 4 is differentiated with respect to I,

\[ \frac{dV}{dln(I)} = n \frac{kT}{q} + I R_s \]
And function $H(I)$ is defined as:

$$H(I) = V - n \frac{kT}{q} \ln \left( \frac{I}{A \exp(T)} \right) = IR_s + n \Phi_B$$

where $I$ is the saturation current. $dV/d\ln(I)$ vs. $I$ and $H(I)$ vs. $I$ plots for the diodes are shown in Figure 7 and electrical parameters that are calculated from these plots are presented in Table I.

Series resistance ($R_s$) provides information about interface properties between graphene and n-Si. Values of the series resistance were calculated from both of the slope of the $dV/d\ln(I)$ vs. $I$ and $H(I)$ vs. $I$ plots and it is decreased with the SAMs surface improvement. In the presence of SAM molecule, incompatible surface between graphene and n-Si is decreased. Therefore, charge transfer is improved owing to due to $\pi-\pi$ interaction, leading to lower resistance at the interface. In addition, while the ideality factor ($n$) that was calculated from the intercept of the $dV/d\ln(I)$ vs. $I$ of the SAM included devices are higher than that of bare graphene diodes, barrier height decreased from 0.931 eV to 0.720 eV with DPIFA and MePIFA surface modification of silicon surface due to compatible interface between graphene and n-Si. Diode with DPIFA SAMs showed better characteristic than that of diode modified by MEPIFA SAMs because, MEPIFA SAMs molecule includes methyl groups restricting $\pi-\pi$ interaction between SAMs and graphene and has longer chain lengths. These limitations of $\pi-\pi$ interaction and extra chain length obstruct charge transfer when compared to DPIFA molecules and thus, diode with DPIFA molecule has lower barrier height and series resistance value (Table I).

### Conclusions

In summary, we investigated electrical characterization of bare and SAMs modified CVD grown graphene/n-Si Schottky diodes. MePIFA and DPIFA molecules were used as SAMs to improve graphene/n-Si substrate interface properties. KPFM technique was performed to define surface potential differences of bare and SAMs modified n-Si. The increments of surface potentials in SAMs modified n-Si cause to reduce energy barrier between aromatic SAMs and graphene with respect to the bare device. Raman spectroscopy and sheet resistance measurement of transferred graphene on SiO$_2$/Si confirmed the presence of the high quality bilayer graphene. The I-V characteristics of bare and SAMs modified graphene/n-type silicon interfaces showed Schottky barrier and rectifying behavior. Additionally, DPIFA SAMs exhibit better diode characteristic compared to MePIFA SAMs due to not containing methyl group which hinders $\pi-\pi$ interaction between SAMs molecule and graphene. Finally, while the ideality factor values increase with modification, barrier height and series resistance values decreased indicating higher diode performance.

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### References


### Table I. Electrical parameters of the Schottky diodes and surface potential values of bare and modified n-Si.

<table>
<thead>
<tr>
<th></th>
<th>$n$ (ideality factor)</th>
<th>$\Phi_B$ (eV) (barrier height)</th>
<th>Surface Potential (V)</th>
<th>$dV/d\ln(I)$</th>
<th>$R_s$ (Ω)</th>
<th>$H(I)$</th>
<th>$R_s$ (Ω)</th>
</tr>
</thead>
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<tr>
<td>n-Si/GRP</td>
<td>1.068</td>
<td>0.931</td>
<td>0.158</td>
<td>301</td>
<td>333.15</td>
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<tr>
<td>n-Si/MePIFA/GRP</td>
<td>1.133</td>
<td>0.820</td>
<td>0.188</td>
<td>208.40</td>
<td>229.86</td>
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<tr>
<td>n-Si/DPIFA/GRP</td>
<td>1.152</td>
<td>0.720</td>
<td>0.383</td>
<td>144.82</td>
<td>169.26</td>
<td></td>
<td></td>
</tr>
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Figure 7. $dV/d\ln(I)$ vs. $I$ and $H(I)$ vs. $I$ plots of the n-Si/GRP/Au, n-Si/MePIFA/GRP/Au and n-Si/DPIFA/GRP/Au diodes.